Benzylic Arylation of 2-Methyl-5 membered Heterocycles Using TMP-Bases

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A new general Pd-catalyzed arylation of various 2-methyl-5-membered heterocycles is reported. This novel method requires $Li-, Mq-,$ or Zn-TMP bases and allows selective metalation of the benzylic position. Subsequent Negishi cross-coupling provides the corresponding arylated heterocycles.

The benzylic arylation of pyridines and related N-heterocycles is of great synthetic interest due to the importance of these heterocycles for pharmaceutical applications.¹ Such arylation reactions on the methylpyridine scaffold often require the preparation of suitable precursors, such as 2-(2 pyridyl)ethanol, 2 N-oxides, 3 or N-iminopyridinium ylides.⁴ Recently, we have shown that a direct benzylic arylation on

picolines, lutidines, and methy-substituted quinolines proceeds readily in the presence of $TMPZnCl \cdot LiCl (1) (TMP =$ 2,2,6,6-tetramethylpiperidyl) and an appropriate Lewis acid such as $\text{Sc}(\text{OTf})_3$ ⁵ However, the arylation of methylsubstituted 5-membered heterocycles remains by far an unsolved problem.6 The arylation of 1,2-dimethylimidazole (2) occurs usually at position 5, and no "benzylic" $C-H$ activation followed by cross-couplings has been reported.^{7,8} (1) (a) Nicolaou, K. C.; Scarpelli, R.; Bollbuck, B.; Werschkun, B.; Only alkylation reactions on the methyl group have been

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Table 1. Pd-Catalyzed Benzylic Arylation of 1,2-Dimethylimidazole (2) Leading to Products of Type 6

performed successfully after deprotonation with n -BuLi.⁹ In addition, the benzylic metalation and subsequent arylation of 2-methylbenzothiophenes¹⁰ and 2-methylbenzofurans^{10b} has not yet been described. Recently, we have developed a range of TMP bases which allow the metalation of various unsaturated molecules under mild conditions.¹¹ Herein, we wish to report that such a direct benzylic arylation can be performed on a variety of different methyl-substituted 5-membered heterocycles, including 2-methylimidazoles, benzothiophenes, and -benzofurans and related 5-membered heterocycles.

Thus, by using the magnesium base $TMPMgCl·LiCl¹²$ (3) and subsequent transmetalation with $ZnCl₂$, we were able to zincate 1,2-dimethylimidazole (2) selectively at the 2-methyl position. The resulting zincated reagent 4 was

used successfully to perform cross-coupling reactions with various aryl bromides of type 5 affording 2-benzylated imidazoles (6) in $71-91\%$ yield (Table 1). For example, the magnesiation of 1,2-dimethylimidazole (2) with $TMPMgCl·LiCl$ (3: 1.5 equiv) in THF is complete within 3 h at 25 °C.¹³ After transmetalation with $ZnCl₂$ (1.5 equiv, 25 °C, 15 min), the zincated imidazole derivative (4) is treated with ethyl 4-bromobenzoate (5a: 0.8 equiv) in the presence of 2% Pd(dba)₂ (dba = dibenzylideneacetone) and 4% SPhos¹⁴ to provide the desired arylated 1,2dimethylimidazole (6a) at 50 °C within 2 h in 85% yield without the formation of a 5-arylated imidazole derivative⁷ (Table 1, entry 1). Similarly, cyano- and trifluoromethylsubstituted aryl bromides (5b,c) can be successfully converted leading to the imidazole derivatives $(6b,c, 71-77\%$, entries 2 and 3). Also, electron-rich aryl bromides bearing alkoxy, pivaloxy, methyl, or amino substituents undergo the cross-coupling in excellent yields $(71-91\%$ yields, entries $4-8$) showing the broad scope of this arylation.

Efforts were made to extend this method to even less acidic methyl-substituted 5-membered heterocycles. We found that 2-methylbenzothiophene (7) was readily metalated with TMPLi¹⁵ (8: 1.15 equiv, -78 °C, 15 min) in THF.¹⁶ After transmetalation with $ZnCl₂$, the resulting heterocyclic benzylic reagent (9) was smoothly arylated with various aryl bromides using 2% Pd(OAc)₂ and 4% SPhos¹⁴ (50 °C, 2 h) leading to the benzothiophenes (10a-h) in $68-98\%$ yield (Table 2, entries 1-8). Interestingly, the 2,3-dimethylbenzothiophene (7b) undergoes an exclusive lithiation at position 2 providing after crosscoupling the 2-benzylated product (10i) in 87% yield (entry 9). A further extension to the 2-methylbenzofuran scaffold was successful using similar conditions. The lithiation of 2-methylbenzofuran (11) was complete within 1 h at -78 °C. After transmetalation with ZnCl₂ and crosscoupling with various aryl bromides, methyl-substituted benzofurans $(13a-d)$ were obtained in $52-75%$ yield (Table 3). In the case of cyano-substituted aryl bromides (5b and 5j), we have found that the addition of 10% Sc(OTf)_{3}^{17} improves the cross-coupling yield.^{5,18}

The generality of our approach is demonstrated by performing the arylation of other related 5-membered heterocycles such as the 2-methylindole derivative 14^{19} and the 2-methylbenzimidazole $15.^{20}$ In these cases, TMPZnCl·LiCl (1) proved to be a suitable base, and a complete zincation could be obtained within 1 h at 25° C. Thus, cross-coupling of 2-methylindole (14) with 4-bromoanisole 5d gave the indole

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Table 2. Benzylic Cross-Coupling of 2-Methylbenzo- $[b]$ thiophene (7)

^a Isolated yield of analytically pure product.

16 in 57% yield (Scheme 1). To the best of our knowledge, this example represents the first benzylic cross-coupling on a 2-methylindole.

In addition, 2-methylbenzimidazole (15) can be arylated successfully after zincation with $TMPZnCl·LiCl$ (1) with 4-bromoanisole (5d) to yield the benzoimidazole derivative 17 in 68% yield (Scheme 1).

In conclusion, we have found a new convenient method for the direct arylation of various 2-methyl 5-membered

(20) For the preparation of the benzimidazole 15, see the Supporting

Table 3. Benzylic Arylation of 2-Methylbenzofuran (11)

^a Isolated yield of analytically pure product. b 10% Sc(OTf)₃ was added. c 2% Pd(dba)₂ and 4% SPhos was used. d 2% Pd(dba)₂ and 2% Xantphos was used.

heterocycles with aryl bromides using standard Pd catalysts. Extension of this method to other 5-membered heterocycles is currently underway.

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Supporting Information Available. Experimental procedures and characterization data of all compounds. This material is available free of charge via the Internet at http://pubs.acs.org.

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The authors declare no competing financial interest.